

Response to Comment on “Determination of *neo*- and *D-chiro*-Inositol Hexakisphosphate in Soils by Solution ^{31}P NMR Spectroscopy”

We welcome the comment from Michael L’Annunziata on our article¹ and his clarification of the debate surrounding his previous identification of *muco*-inositol phosphate in leaf litter of velvet mesquite (*Prosopis velutina*).² We acknowledge our error in ascribing his report to *muco*-inositol hexakisphosphate, when in fact no specific ester form was mentioned in the original article.

We recently analyzed a sample of velvet mesquite litter collected by Dr. L’Annunziata from the same location in Arizona as the sample in his original study. Solution ^{31}P NMR spectroscopy of an alkaline extract of the litter shows no evidence for the presence of inositol hexakisphosphate in any stereoisomeric form (Figure 1a). This includes *muco*-inositol

litter, such signals have to our knowledge not been observed previously in solution ^{31}P NMR spectra of any soil or sediment. This does not preclude the presence of trace amounts of *muco*-inositol hexakisphosphate in some samples, but does support the general assumption that this compound does not occur widely in the environment.

The spectrum of velvet mesquite litter does, however, contain a series of phosphomonoester signals between $\delta = 4.26$ and 5.11 ppm (Figure 1a). While most of these probably originated from the alkaline hydrolysis of RNA and phospholipids,⁴ the presence of a number of simple phosphomonoesters in trace quantities supports Dr. L’Annunziata’s suggestion that the *muco*-inositol phosphate detected in his original analysis was likely to have been a lower-order ester, presumably *muco*-inositol 1-phosphate.

Although more than four decades have passed since publication of Dr. L’Annunziata’s original article,² it remains significant as the only report of a *muco*-inositol phosphate in the environment. Together with the apparent widespread occurrence of the *scyllo*, *D-chiro*, and *neo*-inositol hexakisphosphates in soils and sediments,¹ the presence of *muco*-inositol phosphate in leaf litter provides a further example of the remarkable deficiency in our understanding of the origins, behavior, and function of the inositol phosphate stereoisomers in nature.

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Notes

The authors declare no competing financial interest.

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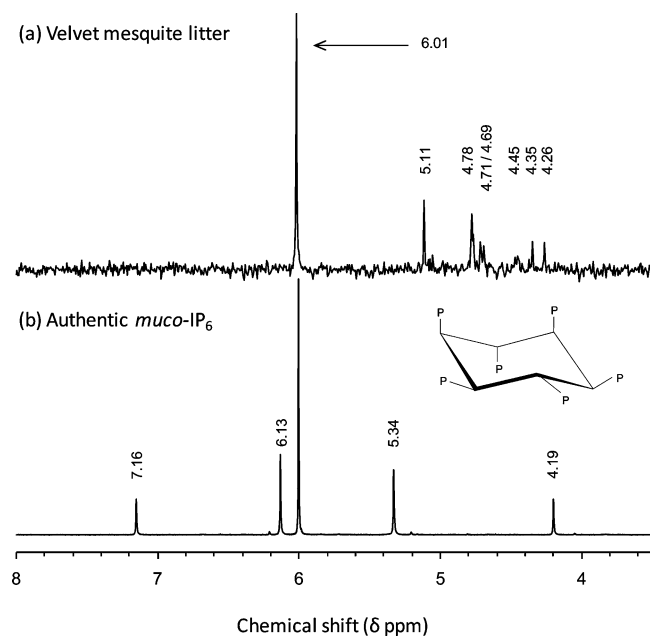


Figure 1. Solution ^{31}P NMR spectra of (a) an NaOH–EDTA extract of velvet mesquite (*Prosopis velutina*) leaf litter from Arizona, and (b) an authentic *muco*-inositol hexakisphosphate (*muco*-IP₆) dissolved in NaOH–EDTA (pH > 13), showing four signals in a 1:2:2:1 ratio. The structure of *muco*-inositol hexakisphosphate is also presented, showing three axial and three equatorial phosphate groups (each group represented by “P”). Chemical shifts of individual signals are given in δ ppm. The strong signal at $\delta = 6.01$ ppm in both spectra is inorganic orthophosphate. Samples were analyzed as described previously.¹

hexakisphosphate, for which we show the spectrum of an authentic preparation of this compound³ from Dennis Cosgrove’s original collection (Figure 1b), kindly provided by Dr Alan Richardson of CSIRO Plant Industry, Canberra, Australia. The authentic *muco*-inositol hexakisphosphate yields four signals in a 1:2:2:1 ratio at $\delta = 4.19$, 5.34, 6.13, and 7.16 ppm (Figure 1b). As well as being absent in the velvet mesquite

(2) L'Annunziata, M. F.; Fuller, W. H. Soil and plant relationships of inositol phosphate stereoisomers: the identification of D-chiro- and muco-inositol phosphates in a desert soil and plant system. *Soil Sci. Soc. Am. Proc.* **1971**, *35*, 587–595.

(3) Cosgrove, D. J. The phosphorylation of *epi*-inositol and *muco*-inositol with polyphosphoric acid. *Carbohydr. Res.* **1975**, *40*, 380–384.

(4) Turner, B. L.; Mahieu, N.; Condon, L. M. Phosphorus-31 nuclear magnetic resonance spectral assignments of phosphorus compounds in soil NaOH–EDTA extracts. *Soil Sci. Soc. Am. J.* **2003**, *67*, 497–510.