Published online 20 January 2013 in Wiley Online Library

(wileyonlinelibrary.com) DOI: 10.1002/oa.2300

Stable Carbon and Oxygen Isotope Spacing Between Bone and Tooth Collagen and Hydroxyapatite in Human Archaeological Remains

C. A. M. FRANCE^{a*} AND D. W. OWSLEY^b

- ^a Smithsonian Museum Conservation Institute, Suitland, MD, USA
- ^b Smithsonian National Museum of Natural History, Washington, DC, USA

ABSTRACT

Spacing between stable isotope values in bones and teeth is a valuable tool for examining dietary influences and diagenesis. This study examines carbon and oxygen isotope values from collagen and hydroxyapatite (structural carbonate and phosphate) in archaeological human bones and teeth to derive species-specific correlation equations and isotope spacing values. The $\delta^{13}C_{\text{collagen}}$ and $\delta^{13}C_{\text{structural carbonate}}$ in bone and dentin collagen show a strong correlation (R = 0.87, 0.90, respectively) with an average $\Delta^{13}C_{\text{carb-coll}}$ spacing of 5.4‰. The consistency of this isotope spacing with other large mammals and in humans with both low and high protein intake (as indicated by enriched $\delta^{15}N$ values) suggests a similar allocation of protein-derived carbon and whole diet-derived carbon to collagen and structural carbonates, respectively, as other terrestrial mammals regardless of absolute meat intake. The $\delta^{18}O_{\text{structural carbonate}}$ and $\delta^{18}O_{\text{phosphate}}$ show the strongest correlation in enamel (R = 0.65), weaker correlations in dentin (R = 0.59) and bone (R = 0.35), with an average $\Delta^{18}O_{\text{carb-phos}}$ of 7.8‰. This isotope spacing is slightly lower than previously reported for large mammals and limited available data for humans. The results potentially indicate species-specific fractionations and differing access to body water and blood-dissolved inorganic carbonates in the presence of collagen formation. The use of correlation between $\delta^{18}O_{\text{structural carbonate}}$ and $\delta^{18}O_{\text{phosphate}}$ to determine diagenetic state is not recommended. The strength of this correlation observed in bones and teeth is variable and alternate indicators of diagenetic state (i.e. C:N ratios of collagen) provide more robust and independent evidence of isotope preservation despite presence/absence of a strong isotope correlation. Published 2012. This article is a U.S. Government work and is in the public domain in the USA.

Key words: bone; carbon isotopes; collagen; hydroxyapatite; human; oxygen isotopes; tooth

Introduction

The use of stable isotopes to examine diet and provenance is common practice in archaeological and paleontological studies. Dietary and regional indications inherent in carbon, oxygen, and nitrogen isotope values of bones and teeth are applied to a myriad of terrestrial vertebrate remains (see reviews in Peterson & Fry, 1987; Ambrose, 1993; Koch *et al.*, 1994; Koch, 1998; MacFadden, 2000). Stable isotope studies of human remains are less common due to the paucity of robust sample sets and the desire to preserve rare or culturally sensitive specimens. Studies are often limited to a specific isotope analysis as opposed to a full suite of isotope indicators, with a few exceptions (lacumin *et al.*, 1996a;

*Correspondence to: Christine A. M. France, Smithsonian Museum Conservation Institute, Suitland, MD, USA.
e-mail: francec@si.edu

Loftus & Sealy, 2012). Understanding of the spacing of isotope values between different components of bones and teeth in humans specifically is limited.

Inferences involving isotope spacing between different tissue and mineral fractions in humans often rely on data from large terrestrial non-human mammals that are more thoroughly studied and better understood. While these data are arguably comparable with humans to some extent, humans have a uniquely omnivorous diet that is often unaccounted for in non-human studies of diettissue isotope spacing with a few exceptions (Krueger & Sullivan, 1984; Lee-Thorp et al., 1989; Hedges, 2003). This study focuses on an unprecedentedly large data set of 18th-19th century human remains to determine accurate human-specific stable isotope spacing values between bone collagen and hydroxyapatite in bones and teeth for individuals reliant on largely terrestrial diets. These data will allow more accurate correlations between isotope values of different tissues and mineral fractions within individuals consuming similar diets, potentially eliminating the need for a full suite of isotope analyses. This study also contributes to understanding pathways between elements in the diet and body tissues and has implications for archaeological diagenetic studies.

Isotope spacing in archaeology

Archaeological bones and teeth contain several components that record isotope information and represent a lifetime average of isotopic input. Collagen is found in both well-preserved bones and tooth dentin. This durable protein contains carbon and nitrogen incorporated directly from the diet (Hedges, 2003). Hydroxyapatite, the inorganic mineral in bones and teeth, contains both phosphate and structural carbonate (-CO₃) substituted in the -PO₄ and -OH locations in this mineral. Both phosphates and structural carbonates preserve oxygen isotopes incorporated through drinking water (Luz & Kolodny, 1985; Bryant & Froelich, 1995; Kohn, 1996). Structural carbonates also preserve carbon isotopes which, similar to collagen, are incorporated from the diet (Hedges, 2003). All isotope values are measured and reported in standard delta notation:

$$\delta X = \left[\left(R_{sample} - R_{standard} \right) / \left(R_{standard} \right) \right] * 1000$$

where X represents the system of interest (i.e. 13 C, 15 N, 18 O) and R represents the ratio of interest (i.e. 13 C/ 12 C, 15 N/ 14 N, 18 O/ 16 O). The carbon standard is Vienna Pee-Dee Belemnite (i.e. V-PDB). The nitrogen standard is atmospheric air. There are two internationally accepted standards for oxygen; this manuscript references all oxygen values to Vienna Standard Mean Ocean Water (i.e. V-SMOW). This study examines the carbon and nitrogen isotopes of bone and tooth collagen (δ^{13} C_{collagen}, δ^{15} N_{collagen}), carbon isotopes of structural carbonate (δ^{13} C_{structural carbonate}), and oxygen isotopes in structural carbonates and phosphates (δ^{18} O_{structural carbonate}, δ^{18} O_{phosphate}). The different carbon and oxygen components are compared for correlations and spacing (Δ^{13} C_{carb-coll}, Δ^{18} O_{carb-phos}).

Carbon isotopes are typically applied to archaeological data sets to determine dietary input due to distinct $\delta^{13}C$ values inherent in plants using the C3 versus C4 photosynthetic pathways (Smith & Epstein, 1971; O'Leary, 1988; Heaton, 1999). Once digested, carbon pathways to the body differ according to the ultimate tissue or mineral into which carbon is incorporated. Previous studies suggest the source of carbon in collagen is mostly dietary protein with lesser contributions from dietary carbohydrates or lipids (Krueger & Sullivan, 1984; Lee-Thorp *et al.*, 1989; Ambrose & Norr, 1993; Tieszen & Fagre, 1993). More recent

research suggests a complex system where consumer collagen carbon isotopes are largely influenced by direct routing of particular amino acids from prey collagen with limited secondary synthesis of others from precursor amino acids or additional dietary components (Howland et al., 2003; Jim et al., 2004; Froehle et al., 2010). This results in an overall diet-collagen fractionation of ~2-5‰ in herbivorous mammals (van der Merwe, 1982; Balasse et al., 1999; Roth & Hobson, 2000; Hedges, 2003; Jim et al., 2004; Warinner & Tuross, 2010; review by Koch, 1998) and ~1-2‰ in carnivorous and omnivorous mammals (Bocherens et al., 1991; Hilderbrand et al., 1996; Bocherens & Drucker, 2003; Coltrain et al., 2004; Fox-Dobbs et al., 2007; Coltrain, 2009). The source of carbon in structural carbonates is blood-dissolved inorganic carbon (DIC) derived primarily from carbohydrates and lipids with a smaller contribution from dietary protein resulting in a δ^{13} C representative of the whole diet (Tieszen & Fagre, 1993; Hedges, 2003; Zazzo et al., 2010). The diet-mineral fractionation for structural carbonates is ~12–15‰ in large herbivorous mammals (Passey et al., 2005; Zazzo et al., 2010; reviews by Koch, 1998; Kohn & Cerling, 2002; Hedges, 2003).

Given these fractionations, the $\Delta^{13}C_{carb-coll}$ observed in mammals is ~5–7% (Krueger & Sullivan, 1984; Lee-Thorp *et al.*, 1989; Ambrose & Norr, 1993; Hedges, 2003; Jim *et al.*, 2004; Warinner & Tuross, 2010), with limited data for humans with a mostly terrestrial diet (lacumin *et al.*, 1996a; Loftus & Sealy, 2012). It should be noted that this average $\Delta^{13}C_{carb-coll}$ value of ~5–7% is often derived from rodents and swine which arguably have a different metabolism, physiology, and diet than humans. Considering that blood- DIC contains contributions from protein, carbohydrates, and lipids, the varying ratio of these contributions in different diets and different metabolisms may produce a species-specific and comparatively unique $\Delta^{13}C_{carb-coll}$ value in humans. Humans in this study are examined for a correlation

Humans in this study are examined for a correlation between $\delta^{13}C_{collagen}$ and $\delta^{13}C_{structural\ carbonate}$ (i.e. variability of $\Delta^{13}C_{carb-coll}$) and potential differences explainable by dietary input. The $\delta^{15}N_{collagen}$ is used as a qualitative indicator of protein consumption. The $\delta^{15}N_{collagen}$ values become more enriched with increased consumption of animal proteins due to the ~3–4% trophic shift inherent in terrestrial food chains (DeNiro & Epstein, 1981; Minagawa & Wada, 1984; Schoeninger & DeNiro, 1984; Sutoh $\it{et}\ al.$, 1987; Post, 2002; Bocherens & Drucker, 2003). A relatively enriched $\delta^{15}N_{collagen}$ value should indicate higher protein input from animal flesh (and therefore higher carbon input from animal flesh) as opposed to dietary input from plants and carbohydrates.

Oxygen isotopes are typically used as proxies for meteoric drinking water in mammals due to the direct correlation between δ^{18} O of local meteoric drinking water, body water, and $\delta^{18}O_{phosphate}$ or $\delta^{18}O_{structural}$ carbonate with subtle variations in fractionation according to species and climate variables (Longinelli, 1984; Luz et al., 1984; Luz & Kolodny, 1985; Levinson et al., 1987; D'Angela & Longinelli, 1990; Bryant & Froelich, 1995; Kohn, 1996; Daux et al., 2008). The relationships between $\delta^{18}O_{phosphate}$, body water, and drinking water in mammals are fairly well understood with several published correlation equations (Longinelli, 1984; Luz et al., 1984; Levinson et al., 1987; Daux et al., 2008). Less well understood is the relationship between $\delta^{18}O_{structural\ carbonate}$ and $\delta^{18}O_{phosphate}$, and relationships between $\delta^{18}O_{structural\ carbonate}$ and body/drinking water. Previously reported fractionation between $\delta^{18}O_{structural\ carbonate}$ and $\delta^{18}O_{phosphate}$ (i.e. $\Delta^{18}O_{carb-phos}$) is ~9–10‰, and fractionation between $\delta^{18}O_{\text{structural carbonate}}$ and body water is ~26.3-27.0% in mammals (Bryant et al., 1996; Iacumin et al., 1996b; Martin et al., 2008; Pellegrini et al., 2011), with limited data for humans specifically (Iacumin et al., 1996a). This study examines the $\delta^{18}O$ correlation between phosphate and structural carbonate in an effort to accurately predict $\delta^{18}O_{phosphate}$ values from $\delta^{18}O_{\text{structural carbonate}}$. The former require more expensive offline chemical purifications, and the most expedient mass spectrometry methods to analyze δ¹⁸O_{phosphate} typically show poorer reproducibility (see Methods). It is tempting to streamline the chemical and mass spectrometry procedures and use the better understood phosphate equations to obtain $\delta^{18} O_{phosphate}$ from $\delta^{18} O_{structural\ carbonate}.$ However, this calculation cannot be performed without observed consistency in $\Delta^{18}O_{carb-phos}$ and a value specific for

This study analyzes a set of 18th–19th century North American archaeological remains. Individuals from this time period had a more localized diet than modern humans thus eliminating variability due to global dietary influences inherent in most modern diets. All

individuals resided primarily in the eastern United States which shows a limited range in the baseline $\delta^{15}N$ values of vegetation of approximately -4.0 to 0.0% with a few isolated exceptions (Handley *et al.*, 1999; Nadelhoffer *et al.*, 2004; Billings & Richter, 2006; Pardo *et al.*, 2007; Templer *et al.*, 2007). These similar and localized diets allow the use of $\delta^{15}N$ values as independent measures of protein input as opposed to an indication of provenance. This sample set is also relatively recent which greatly reduces the likelihood of post-mortem diagenetic alteration, although all samples are individually examined for diagenetic alteration in the course of study.

Methods and materials

Samples

Samples consist of 18th–19th century North American human remains. Archaeological sites are located primarily in the mid- and north-eastern United States with two exceptions in New Mexico (Table 1). All individuals resided in the eastern United States although several are buried in the west due to death during military service.

Chemical extraction

Specimens were exhumed, mechanically cleaned, documented, and catalogued. All specimens were stored without chemical treatments or bone consolidants. Samples were prepared mechanically using a common rotary tool. Whole bone plugs or tooth roots were separated for collagen analyses. Bone, dentin, and enamel were crushed to a coarse powder for phosphate and structural carbonate analyses.

Bone and dentin collagen was extracted according to modified methods of Longin (1971), DeNiro & Epstein (1978), and Bocherens *et al.* (1991). Whole bone or tooth roots (~200 mg) were sonicated in ultra-pure water and

Table 1. Sample sites, locality, and time period

Site	Site Location	Time period	n
Congressional Cemetery	District of Columbia	~1850–1900	34
Trinity Catholic Church	District of Columbia	~1800–1850	23
Woodville Cemetery	Delaware	1790–1850	10
Walton Family Cemetery	Connecticut	~1750–1830	20
Glorieta Pass	New Mexico	1862	38
First African Baptist Church	Pennsylvania	1824–1842	9
Parkway Gravel	Delaware	1800–1900	5
Fort Craig	New Mexico	1854–1877	58

rinsed to remove extraneous dirt and labile salts. Samples were decalcified in 0.6 M hydrochloric acid at 4°C for several days, with fresh acid added daily, until the reaction completed; average reaction time was 3–5 days. After rinsing in ultra-pure water, the remaining crude protein was soaked in 0.125 M sodium hydroxide for 18 h to remove humic and fulvic acids. Samples were rinsed and reacted in 0.03 M hydrochloric acid at 95°C for 18 h to separate hot water soluble and insoluble phases. The resulting supernatant was lyophilized to extract purified collagen.

Phosphate in the hydroxyapatite was extracted according to the method of Dettman *et al.* (2001). Bone and tooth powders (~20 mg) were soaked in 2 M hydrofluoric acid solution overnight to liberate phosphate ions. The solution was diluted and buffered with 20% ammonium hydroxide before adding 2 M silver nitrate. The resulting silver phosphate precipitate was rinsed with ultra-pure water and dried (60°C).

Structural carbonate in the hydroxyapatite was extracted according to modified methods of Bryant et al. (1996). Bone and tooth powders (~20 mg) were soaked in 2–3% sodium hypochlorite overnight to remove organic substances. After rinsing in ultra-pure water, samples were soaked in 1 M acetic acid solution buffered with 1 M calcium acetate (pH ~ 4.5) for 4 h to remove secondary carbonate phases. The remaining material was rinsed in ultra-pure water and dried (60°C) .

Mass spectrometry methods

All analyses were conducted on Thermo Delta V mass spectrometers at the Smithsonian OUSS/MCI Stable Isotope Mass Spectrometry Laboratory. Collagen was weighed into tin capsules (~0.5 mg) and combusted in a Costech 4010 elemental analyzer (EA) producing CO₂ and N₂ gases which were introduced to the mass spectrometer via a Conflo IV interface and measured for $\delta^{13}C_{\text{collagen}}$ and $\delta^{15}N_{\text{collagen}}$. Silver phosphates were weighed into silver capsules (~0.5 mg) and thermally decomposed in a Thermo temperature conversion EA to CO gas which was introduced to the mass spectrometer via a Conflo IV interface and measured for $\delta^{18}O_{phosphate}$. Structural carbonate samples were acidified in 102% phosphoric acid (density ≥1.92) at 25°C for 18 h in a pure helium environment producing CO₂ gas which was introduced to the mass spectrometer via a Gas Bench II system and measured for $\delta^{13}C_{structural\ carbonate}$ and $\delta^{18}O_{structural\ carbonate}$.

Collagen samples were linearly corrected to a house acetanilide standard and the urea UIN-3 standard (Schimmelmann *et al.*, 2009), both of which are

calibrated to international standards USGS-40 and USGS-41. All $\delta^{13}C_{collagen}$ and $\delta^{15}N_{collagen}$ data are reported with the error inherent in the international standards used for calibration (i.e. $\pm 0.2\%$, 1σ), internal reproducibility was <0.2% (1 σ). Structural carbonate samples were linearly corrected to the international standards NBS-19 and LSVEC. All $\delta^{13}C_{structural\ carbonate}$ and $\delta^{18}O_{structural\ carbonate}$ data are reported with the error inherent in these standards (i.e. $\pm 0.2\%$, 1σ), internal reproducibility was <0.2% (1 σ). Phosphate samples were linearly corrected to the international standards USGS-34 and USGS-35. The error inherent in these standards is $\pm 0.2\%$ (1 σ), internal reproducibility was <0.4% (1 σ). All $\delta^{18}O_{phosphate}$ data are therefore reported with an error of $\pm 0.4\%$ (1 σ).

Statistical analyses

All regressions are parametric least squares regressions assuming normal distribution and constant variance. All correlations are Pearson product moment correlations assuming normal distribution and constant variance of the residuals.

Results

Diagenesis

Although the samples are relatively modern, their burial introduces potential for post-mortem diagenetic alteration. Data were selected for inclusion in analyses based upon the previously established C:N ratio range of 2.8-3.6 and a weight %N yield of ~11-16% for well-preserved collagen (DeNiro, 1985; Ambrose, 1990; Bocherens et al., 1991, 1994, 1996, 1997; Drucker et al., 2001, 2003; McNulty et al., 2002; Coltrain et al., 2004; Jorkov et al., 2007). Approximately 80% (154 of 197) of analyzed samples satisfied these requirements (Figure 1). While criteria for the preservation of original phosphate and structural carbonate isotope values has been considered (Tuross et al., 1989; Michel et al., 1995; Person et al., 1995, 1996; Iacumin et al., 1996b; Kohn et al., 1999; Zazzo et al., 2004), this study includes only those samples yielding wellpreserved organic collagen per the above stated criteria. It has been noted that mineral crystals in bones and teeth are nested within the organic matrix (Francillon-Vieillot et al., 1990; Veis, 2003), such that the presence of wellpreserved organic matter protects the mineral fraction of bones and teeth from recrystallization and subsequent isotope alteration (Nelson et al., 1986; Person et al., 1996; Tütken et al., 2008).

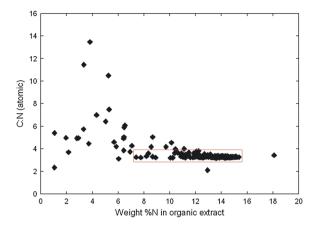


Figure 1. C:N ratios and weight %N yields from organic extracts. The rectangle identifies the samples that meet the criteria for good preservation.

Stable isotope ranges and correlations

The $\delta^{13}C_{collagen}$ ranges from -20.9 to -7.5%; the $\delta^{13}C_{structural\ carbonate}$ ranges from -14.9 to -5.2% (Table 2). The $\delta^{13}C_{collagen}$ and $\delta^{13}C_{structural\ carbonate}$ are strongly correlated (Figure 2, R = 0.88). Both dentin and bone exhibit similar trends with strong correlations (R = 0.90, 0.87, respectively). Basic linear regression relationships are listed in Table 3. The average $\Delta^{13}C_{carb-coll}$ is $5.4\pm1.1\%$ (1 σ) with little variation between dentin and bone. No significant correlation exists between $\Delta^{13}C_{carb-coll}$ and $\delta^{15}N_{collagen}$ (Figure 3, R = 0.26).

The $\delta^{18} O_{structural\ carbonate}$ ranges from +20.4 to +30.9%; the $\delta^{18} O_{phosphate}$ ranges from +13.3 to +22.8% (Table 2). The correlation between $\delta^{18} O_{structural\ carbonate}$ and $\delta^{18} O_{phosphate}$ is moderate (Figure 4, R = 0.67). Tooth enamel shows the strongest correlation between values (R = 0.65), while dentin and bone show weaker correlations (R = 0.59, 0.35, respectively). Basic linear regression relationships are listed in Table 3. The average $\Delta^{18} O_{carb\text{-}phos}$ is 7.8 \pm 1.5% (1 σ) with little variation between dentin, enamel, and bone. Enamel shows a notably lower variability (\pm 1.0, 1 σ) compared to dentin (\pm 1.5, 1 σ) and bone (\pm 1.6, 1 σ).

Note that isotope data is not distinguished by site locality. While absolute values of $\delta^{13}C_{collagen}$, $\delta^{13}C_{structural}$ carbonate, $\delta^{18}O_{structural}$ carbonate, and $\delta^{18}O_{phosphate}$ differ by locality due to differing regional vegetation and meteoric water composition, there were no observable trends between $\Delta^{13}C_{carb-coll}$ or $\Delta^{18}O_{carb-phos}$ and locality. Likewise, $\delta^{15}N_{collagen}$ shows no correlation with individual sites. The implications of regional differences in isotope values will be discussed in a future manuscript.

Discussion

The correlation between $\delta^{13}C_{collagen}$ and $\delta^{13}C_{structural}$ carbonate is robust with fairly consistent $\Delta^{13}C_{carb-coll}$ values of ~4.2–6.4‰. Previous studies of humans show Δ^{13} C_{carb-coll} values of ~2.6–6.2‰, the lower range of which is found in individuals with a partially marine diet (Iacumin et al., 1996a; Loftus & Sealy, 2012). Data from this study originates from individuals with strictly terrestrial diets and agrees with the larger spacing observed in similar individuals (lacumin et al., 1996a), as well as previous determinations of the offset for omnivorous non-human mammals (Krueger & Sullivan, 1984; Lee-Thorp et al., 1989; Hedges, 2003). This observation suggests that carbon diet-tissue fractionations between different organic and mineral fractions of bone and dentin function similarly in humans and other large mammals. The strong correlation supports consistent fractionation of carbon isotopes between diet and carbon-containing fractions (i.e. collagen or hydroxyapatite) in bones and teeth. The $\Delta^{13}C_{carb-coll}$ does not correlate with $\delta^{15} N_{collagen}$ which suggests that diet in these humans is balanced such that the ultimate source of collagen carbon is consistently protein from animal flesh while the contribution to hydroxyapatite contains components from the whole diet. While the $\delta^{15}N_{collagen}$ serves as a rough indicator of the relative meat proportion in the diet, it appears that individuals with relatively lower animal protein intake are still allocating a similar proportion of animal derived carbon (compared to carbohydrate or lipid derived carbon) to their collagen.

The average $\Delta^{18} O_{carb\text{-phos}}$ of 7.8 \pm 1.5% (1 σ) is less than previously observed values of ~9-10% (Bryant et al., 1996; Iacumin et al., 1996a, 1996b; Martin et al., 2008; Pellegrini et al., 2011). While $\Delta^{18}O_{carb-phos}$ variation between dentin, enamel, and bone is minimal, enamel and dentin show notably lower intra-variability. Correlation between $\delta^{18}O_{\text{structural carbonate}}$ and $\delta^{18}O_{\text{phosphate}}$ values is only moderate at best with the strongest correlation noted for tooth enamel. All data included in these calculations demonstrated good collagen quality, thereby rendering diagenetic alteration an unlikely explanation for variability observed in these correlations or the $\Delta^{18}O_{carb-phos}$. This observation suggests that fractionation of oxygen when incorporated into -CO₃ or -PO₄ differs according to the mineral and body element. It is typically assumed that both the carbonate and phosphate precipitate from the same body water pool (Bryant & Froelich, 1995; Kohn, 1996); the resulting $\delta^{18}O$ values should therefore correlate. Data from human enamel in this study support this idea, but data from bone and dentin show support to a lesser degree.

Table 2. Data. Only data derived from samples determined to be well-preserved and therefore used in calculations are listed here

:	ī	δ ¹⁸ O _{phosphate}	δ ¹⁸ O _{structural} carbonate	Δ ¹⁸ O _{carb-phos}	δ ¹³ Cstructural carbonate	δ ¹³ C _{collagen}	Δ ¹³ C _{carb-coll}	8 ¹⁵ N _{collagen}	N S S
Designation	Element	(%o, VSMOW)	(‰, VSMOW)	(‰, VSMOW)	(‰, VPDB)	(‰, VPDB)	(‰, VPDB)	(‰, air)	
291 A1091-BOB-05	Metatarsal	17.9	26.5	α Ω	-102	-15.9	5.7	10.4	C.
29LA1091-BOR-06	Metatarsal	18.5	27.1	8.7	-11.2	-17.4	6.2	6.6	3.5
29LA1091-BOR-07	Metatarsal	17.4	27.1	9.7	-10.1	-17.2	7.1	12.3	ლ ლ
29LA1091-BOR-11	Metatarsal	D: / L	27.5	~ C	0.7–	0.11.6	7.4	- c	ю с И с
29LA1091-BOR-13	Metatarsal	4.01	- 55 - 85 - 86	, с с с	0.00	10.0	~ ч ч		3 7 9
29LA1091-BOR-14	Metataroal	17.1	23.0 27.5	0.0	-10.9 -7.5	1.7.1	0.0	7 - 1	ე (ე (
291 A1091-BOB-18	Metatarsal	÷ 00	26.75	- 6	-6.7	12.4.7	. v.	13.1	1 (C
291 A1091-BOR-19	Metatarsal	17.0	27.1	10.1	-6.0	120	9 9	10.3) (C
29LA1091-BOR-23A	Talus	17.5	26.3	- œ	0.60	15.1	9.0	1.5	1 (C)
29LA1091-BOR-24	Metatarsal	13.3	23.2	6.6	6.9	-12.3	5.4	10.5	3.5
29LA1091-BOR-26	Metatarsal	17.4	23.9	6.5	-5.7	-10.9	5.2	12.0	3.4
29LA1091-BOR-30	Metacarpal	15.8	24.5	8.7	-11.5	-18.1	9.9	11.3	3.3
29LA1091-BOR-31	Metatarsal	16.2	24.6	8. 4.0	-5.2 1.5	11.	ල. ග.	11.5	ლ ლ
29LA1091-BOR-32	Metatarsal	E 7	24.4		8.7-	- 1 3. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5. 5.	υ ΩΩ	0.0	2) (4 (
29LA1091-BOR-33	Metatarsal	- u	0.450	ю с 4. с	4.0.1	10.1	7.7	0. 0.	ა ი ა ი
29LA1091-BOR-34 29LA1091-BOB-35	Metatarsal	0.00	24.0	თ დ ი თ	-6.1	11.07	, r.	10.7	0, 60 1/ 10
29LA1091-BOR-38	Metatarsal	17.2	29.7	12.55	7.7	-14.0) (C)	0	0 0
29LA1091-BOR-39	Metatarsal	16.6	22.5	0.00	-10.3	-15.3	5.0	11.9	3.6
29LA1091-BOR-39	Metatarsal	16.6	20.8	4.2	-10.0	-15.0	5.0	12.3	3.3
29LA1091-BOR-39	Metatarsal	16.1	22.5	6.4	-10.3	-15.3	5.0	11.9	დ (
29LA1091-BOR-40	Metacarpal	5.7	24.6	ა. ე. ი	-5.4 4.0	-11.6	6.1	10.7	ა. 4. ი
29LA1091-BOR-41	Mototopi	- C	20.00 0.00 0.00	о ц о т	0.07	0.00	о о г	v + v ∠	
29LA 109 1-BOR-43	Metatarsal	5.0 0.0	26.3	. w	7. 6.	 	. 7.	t 0:	ე დ 1 4
29LA1091-BOR-45	Metatarsal	18.5	24.1	5.6	-11.3	-17.6	. e. e.	0.11	9.8
29LA1091-BOR-47	Metatarsal	17.5	23.3	5.8	6.8	-14.3	5.5	6.6	3.3
29LA1091-BOR-49	Metatarsal	17.0	25.9	8.9	-10.0	-16.7	6.7	11.4	3.2
29LA1091-BOR-50	Metatarsal	16.6	24.5	0.0	-10.0	-17.3	4.7	0.0	ლ ლ ი
29LA1091-BOR-51	Metatarsal	7.07	Z0.02	ر بن د	1.1-	 3 3 3 3	Ω. α Σ. τ	0.0	ນ (ນ (
29LA 1091-BOR-34 291 A 1091-BOR-55	Metatarsal	17.7	24.3	7 0.0	0.71	1 1 0 0 0 0	- m	10.5	ე თ ე თ
29LA1091-BOR-57	Metatarsal	19.2	25.1	5.0	-5.6	-11.4	5.9	10.4	3.5
29LA1091-BOR-58	Metatarsal	17.2	23.7	6.4	-12.0	-18.6	9.9	11.3	3.4
29LA1091-BOR-62	Metatarsal	18.7	21.9	3.2	-7.0	-12.2	5.2	11.3	3.3
29LA1091-BOR-64	Metatarsal	16.2	25.1	o o	12.3	-19.2	0.7	0. F	() ()
ZSLA 109 1-BOR-63	Metatarsal	0.0	7.02.0	\ O \	12.7	1.0.7). ()	- T	о с И с
28LA 1091-BOR-60 510A ISTEN-00-07	Metacarpal	4- α υ	20.00 20.000 20.0000	ກ ດ ວ ຜ	5.01- 5.01-		- r. л а	φ. «	η α
5107001EL-00-07	Metacarpal	- 6	21.5	o rc	-117	-17.0	or co	- - -) (C
51CAUSTEN-CC-11	Metacarpal	16.9	21.5	4.6	-12.2	-16.7	4.5	10.9	3.6
51CAUSTEN-CC-12	Metacarpal	17.3	24.0	9.9	-11.8	-16.2	4.4	12.1	3.2
510AUSTEN-00-13	Metacarpal	15.7	20.8	5.0	-10.9	-16.1		11.7	დ დ (
STCAUSTEN-CC-14	Metacarpal	4.7.	23.4	0.0	U 0	7.7.1	υ. Σ	- o	ω c ω <
51KFYWORTH-CC-0-	Metatarsal	0.07	24.4 24.3	o. %	0.00	150	4. ₪ 	10.7	ე დ 4 დ
)) - - -	;	9))	:)

$\begin{array}{cccccccccccccccccccccccccccccccccccc$	
1100 0 0 0 0 1 1 1 1 1 1 1 1 1 1 1 1 1	
$\begin{matrix} -4 & 0 & 0 & 0 & 0 & 0 & 0 & 0 & 0 & 0 & $	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	7
	-
$\begin{array}{llll} \mathcal{L}\otimes\mathcal{L} & \mathcal{L}\otimes\mathcal{L}\otimes\mathcal{L}\otimes\mathcal{L}\otimes\mathcal{L}\otimes\mathcal{L}\otimes\mathcal{L}\otimes\mathcal{L}\otimes$	
48888848488888888888888888888888888888	
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	<u>.</u>
Radius Fibula Metatarsal Tibia Metatarsal Tibia Metatarsal Temporal Femur Metacarpal Fibula Radius Fibula Radius Fibula Radius Femur	
51KEYWORTH-CC-04 51KEYWORTH-CC-05 51KEYWORTH-CC-08 51KEYWORTH-CC-08 51KEYWORTH-CC-10 51KEYWORTH-CC-12 51KEYWORTH-CC-13 51KEYWORTH-CC-14 51KEYWORTH-CC-14 51KEYWORTH-CC-14 51KEYWORTH-CC-14 51KEYWORTH-CC-14 51KEYWORTH-CC-13 51KHITE-CC-03 51WHITE-CC-03 51WHITE-CC-03 51WHITE-CC-03 51WHITE-CC-03 51WHITE-CC-03 51WHITE-CC-03 6CT58-5-AMM03 6CT58-5-AMM04 6CT58-5-AMM05 6CT58-5-AMM05 6CT58-5-AMM27 6CT58-5-AMM27 6CT58-5-AMM26 6CT58-5-AMM27 6CT58-5-AMM26 6CT58-5-AMM27 6CT58-5-AMM27 6CT58-5-AMM26 6CT58-5-AMM27 6CT58-5-AMM26 6CT58-5-AMM27 6CT58-5-AMM27 6CT58-5-AMM27 6CT58-5-AMM27 6CT58-5-AMM27 6CT58-5-AMM27 6CT58-5-AMM27 6CT58-5-AMM27 6CT58-5-AMM27 6CT58-5-AMM27 6CT58-5-AMM27 7NCE176-DHCA-V03 7NCE176-DHCA-V04 7NCE176-DHCA-V04 7NCE176-DHCA-V04	SLOPEB

Table 2. (Continued)									
: :+ :-		8 ¹⁸ O _{phosphate}	δ ¹⁸ O _{structural} carbonate	Δ^{18} Ocarb-phos	8 ¹³ Cstructural carbonate	$\delta^{13} C_{collagen}$	$\Delta^{13}C_{\text{carb-coll}}$	$\delta^{15} N_{collagen}$	Z. Ö
Designation	Element	(‰, VSMOW)	(‰, VSMOW)	(‰, VSMOW)	(‰, VPDB)	(‰, VPDB)	(‰, VPDB)	(‰, air)	
7NCE98A-DHCA-01	Metacarpal	17.8	25.6	7.8	-6.1	11.3	5.2	11.2	
/NCE36A-DHCA-06 FABC-08-107a	Metacarbal	16.2 5.2	24.3 2.8	0 8 4 0	-10.6	- 10.3 - 16.2	4 rv 7 0.	10.3	
FABC-08-3300	Metatarsal	16.7	24.2	7.4	-11.5	-17.0	5.5	8.8	
FABC-08-63b	Metatarsal	16.8	23.8	7.1	11.1	-16.6	5.5	10.5	
FABC-08-8000	Metacarpal	16.2	23.5	/ L	8.7	12.4	4. 3.	70.7	
74BC-08-9100 GI 0-099-1A	Fnalanx	- 1 - 7.0 - 7.0	24.1 27.0	- v 5	- 15.00 - 7.51	1 1 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7 7	о г. Б	- 1 - 5	
GLO-099-2A	Femur	17.6	26.7) () ()	ာ တ	15.2		10.8	
GLO-099-2C	Femur	18.6	26.0	4.7	1 0 0	-12.3	5.4	10.4	3.3
GLO-099-2E	Femur	19.2	26.5	7.3	7.0	-12.9	5.0 0.0	10.4	
GLO-089-2H GLO-099-2R	remur Metatarsal		27.22	8. 4	3.7 3.00 6.00	17.55	5.0	10.0 V	
GLO-099-2 V	Femur	18.9	26.8	7.9	-8.7	-12.5	. 8. 8.	10.3	
GLO-099-2X	Femur	16.9	25.1	8.2	-10.6	-12.7	2.1	10.1	
TRINITY-EAST-10	Metacarpal	17.5	24.5 26.5	7.0	7.5	12.2	6.4 6.6	1 1 8 1 8	
TBINITY-FAST-113	Metacarnal	ا ش د	20.2 20.2 20.2	ς α	0. 0.	0.1	2. A	1 - 1 2 - 1 - 1	
TRINITY-EAST-14	Temporal		25.8))	0.7-	-12.3 	. 4	10.4	
TRINITY-EAST-26	Temporal	18.2	25.6	7.4	-8.4	-12.9	4.5	11.3	
TRINITY-WEST-01	Temporal	16.6	24.6	0.0	8.2	-10.8	2.5	9.7 0.0	
29LA1091-BON-00 29LA1091-BOR-06	Premolar - denuir		20.5	0 O	14.7	0.02-	4.0	V. 0	
29LA1091-BOR-07	Premolar - dentin		22.4	4.6	-12.1	-19.2	7.1	12.1	3.5
29LA1091-BOR-07	Premolar - ename		25.4	5.4	-12.3	1	1	1	
29LA1091-BOR-13	Premolar - dentin		26.8	ත I	-10.9	-16.6	2.7	1.1	3.3
29LA1091-BOR-14	Premolar - ename		25.6 3.0 3.0	/:/	-13.0	I	I	I	I
29[A1091-BON-15	Premolar - dentin		26.5	. 60 . 7. 7.	† 6: 	-13.0	9	10.3	(2)
29LA1091-BOR-47	Premolar - enamel		26.3	8.7	-6.4))	
29LA1091-BOR-49	Molar - dentin		24.8	6.7	-13.0	-20.7	7.8	11.4	3.6
29LA1091-BOR-49	Molar - enamel		26.6	න ග ග	-14.2 0.5	I	I	I	ı
51KEYWOKIH-CC-05 51KEYWORTH-CC-15	Premolar - enamel Premolar - dentin		2,02,0 2,03,00	0 0 0) (2) (3) (4) (5)	17_8	<u> </u>	100	رب ۱ رد
51KEYWORTH-CC-12A	Premolar - enamel	_	26.3	9.0	- 1 5 65 5 65	2) F	7 -))
7NCE176-DHCA-X01L	Molar - dentin		24.5	9.9	7.7—	-12.8	5.1	10.6	3.4
GLO-099-2A	Molar - dentin	16.7	27.4	10.7	_9.2 2.6	-15.5	6.3	10.8	3.3
GLO-099-2A	Molar - enamel	16.9	26.4	დ დ. დ	ა. ე.	1 0	1 1	1 (1 0
GLO-099-ZAA	Canine - dentin	20.0 V V	28.20 4.00 4.00	8. V 4. Q	∞	72.5	υ 4.	70.7	ω I
GLO-099-2B	Canine - dentin	t	28.7	. I	0.00	-11.9	5.1	10.1	33
GLO-099-2BB	Molar - dentin	21.1	28.3	7.3	-8.1	-13.7	5.6	12.1	3.3
GLO-099-2BB	Molar - enamel	72.3 10.8	28.9	6.0 0.0	4.7– o 5	1 0	<	1 0	0
GLO-099-2C	Molar - enamel	20.4	27.4	7.0	- 1 - 8 - 1	J 5 1	t f	<u>.</u> 1) I
GLO-099-2CC	Molar - dentin	16.8	26.3	9.0	-12.9	-18.2	5.4	8.1	3.1
GLO-088-700 GLO-088-700	Molar - enamel Canine - dentin	0 0 0 0	.30.5 .30.8.	20.01 7.7.	13.0 6.4	-127	с: I С	101	(°.
])	-)))

Published 2012. This article is a U.S. Government work and is in the public domain in the USA. Int. J. Osteoarchaeol. 25: 299–312 (2015)

GLO-099-2DD GLO-099-2EE	Canine - enamel Canine - dentin	20.4	27.4 29.2 30.5	7.0	-6.7 -5.9	10.3	4.4	11.0	3.2
GLO-099-2EE GLO-099-2G	Canine - enamel Canine - dentin	17.7	26.6 20.6	- 0. c	1.0.1 4.0.1	-19.7		10.9	
GLO-099-2 G GLO-099-2H	Canine - enamel Molar - dentin Molar	20 c 20 c 20 c 20 c 20 c 20 c 20 c 20 c	28.7 28.7 20.4	ω ω υ 4 τ		_ -12.6		11.0	
GLO-039-2H GLO-039-2I	Molar - erlarnel Canine - dentin	20.0	28.4 4.45	 	-0.6 -7.4	-12.2		10.0	3
GLO-099-21	Canine - enamel	19.6	28.4	8.8	-7.4	! !			!
GLO-099-2J	Canine - dentin	19.2	28.8	9.7	-7.4	-12.5			3.3
GLO-099-2J	Canine - enamel	20.2	28.3	 		(T			1 0
GLO-088-7.5 GLO-098-2.5	Molar - dentin Molar - enamel	20.5 21.5	- 00 00 00 00 00 00 00 00 00 00 00 00 00	7.50	7.0	0.			ا ر _د در
GLO-099-2 L	Canine - dentin	17.9	26.8	0.00	-11.8	-15.8			3.2
GLO-099-2 L	Canine - enamel	19.2	26.4	7.2	-12.3	ı			ı
GLO-099-2 M	Molar - dentin	19.0	28.9	6.6	-5.7	-12.1			3.3
GLO-099-2M	Molar - enamel	19.6	28.1	8.5	-5.7	L			1
GLO-099-2 N	Molar - dentin	19.9	28.1	0.00 7.7	- 00.0	-13.5			3.3
GLO-099-2N	Molar - enamel	19.5	27.6	8.1	-9.1	1 :			
GLO-099-20	Molar - dentin	20.9	29.2	က်လ	1.6	-11.6 0.6			ლ ლ დ
GLO-099-2P	Canine - dentin	20.3	28.9	9.0	7.2	-9.2			(N)
GLO-099-2F	Canine - enamel		78.1	%.v	4.7	1 0			1 0
GLO-099-2Q	Molar - dentin	18.7	28.5	ω. c) XX XX +	13.3			χ, c
GLC-099-25	Molar - dentin		28.6	10.5	- L - C - C	- 14.0 - 14.0			v v
GLO-099-2S	Molar - enamel	21.6	30.1	9.8	9.6) -)))
GLO-099-2T	Canine - dentin	19.9	29.6	9.7	-7.5	-15.0			3.5
GLO-099-2T	Canine - enamel	20.1	29.6	9.6	-6.7	I			ı
GLO-099-2U	Canine - dentin	16.7	26.8	10.2	-12.0	-18.8			3.2
GLO-099-2U	Canine - enamel	18.6	26.3	7.7	-12.4	ı			1
GLO-099-2 V	Molar - dentin	19.0	30.1	-	-6.7	-12.7			3.3
GLO-099-2V	Molar - enamel	21.1	28.8	7.6	-5.8	ı			ı
GLO-099-2X	Incisor - dentin	I	28.1	ı	-7.8	-13.9			3.5
GLO-099-2Y	Molar - dentin	17.1	26.5	9.5	-10.7	-10.7			3.3
GLO-099-2Y	Molar - enamel	18.2	25.3	7.1	-13.2	ı			1
GLO-099-2Z	Canine - dentin	1	28.5	1	-6.3	-12.1			3.2
TRINITY-EAST-22A	Molar - dentin	19.6	27.0	7.4	-14.6	-20.9			(C)
	Molar - enamel	21.0	28.5	7.5	-15.0	1 (L
I KINITY-EAST-25	Premolar - dentin	I	26.9	I	-14.9	-20.8			3.5
									I

Published 2012. This article is a U.S. Government work and is in the public domain in the USA. Int. J. Osteoarchaeol. 25: 299–312 (2015)

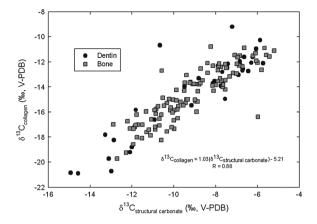


Figure 2. Carbon isotope values from structural carbonate and collagen. The displayed formula is a linear regression on all data.

It is however known that the carbonate fraction of hydroxyapatite also incorporates an isotope contribution from blood- DIC. This same pool of blood-DIC is tapped during collagen formation, albeit to a much lesser degree (Jim et al., 2004; Froehle et al., 2010). It is possible that the pool of blood-DIC is therefore fractionated during collagen formation, which occurs before mineralization of hydroxyapatite (Francillon-Vieillot et al., 1990; Veis, 2003), thus diminishing the correlation between the $\delta^{18}{\rm O}_{\rm structural\ carbonate}$ and $\delta^{18}{\rm O}_{\rm phosphate}$ values in bone and dentin. The noted difference between the $\Delta^{18}{\rm O}_{\rm carb-phos}$ value of 7.8‰ in humans from this study and values of ~9–10‰ observed in other research focusing on non-human mammals could also be due to a species-specific metabolic or physiological mechanism.

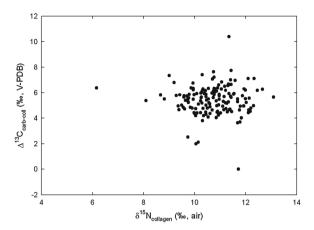


Figure 3. Nitrogen isotope data from collagen and the spacing between carbon isotopes in structural carbonates and collagen.

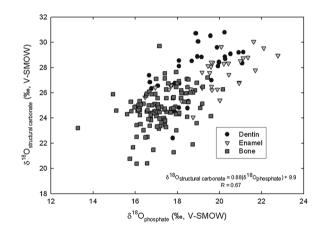


Figure 4. Oxygen isotope values of structural carbonates and phosphates. The displayed formula is a linear regression on all data.

Table 3. Linear regression equations

Carbon isotope comparisons. All formulas based on $\delta^{13}\text{C}$ referenced to V-PDB.

Body Element	Regression Formula	R	Average Δ ¹³ C _{carb-coll}
Bone and dentin combined Bone Dentin	$\begin{array}{l} \delta_{^{13}C_{collagen}}^{13} = 1.03 (\delta_{^{13}C_{structural\ carbonate}}^{13}) - 5.21 \\ \delta_{^{13}C_{collagen}}^{13} = 1.00 (\delta_{^{13}C_{structural\ carbonate}}^{13}) - 5.48 \\ \delta_{^{13}C_{collagen}}^{13} = 1.07 (\delta_{^{13}C_{structural\ carbonate}}^{13}) - 4.78 \end{array}$	0.88 0.87 0.9	5.4 ± 1.1‰ (1σ)

Oxygen isotope comparisons. All formulas based on δ^{18} O referenced to V-SMOW.

Body Element	Regression Formula	R	Average Δ ¹⁸ O _{carb-phos}
Bone, dentin, enamel combined Bone Dentin Enamel	$\begin{array}{l} \delta^{18}O_{structural\ carbonate} = 0.88(\delta^{18}O_{phosphate}) + 9.9\\ \delta^{18}O_{structural\ carbonate} = 0.51(\delta^{18}O_{phosphate}) + 15.9\\ \delta^{18}O_{structural\ carbonate} = 0.87(\delta^{18}O_{phosphate}) + 11.3\\ \delta^{18}O_{structural\ carbonate} = 0.63(\delta^{18}O_{phosphate}) + 14.9 \end{array}$	0.67 0.35 0.59 0.65	7.8 ± 1.5‰ (1σ)

With only moderate correlation between $\delta^{18}O_{structural}$ carbonate and $\delta^{18}O_{phosphate}$, the use of one to predict the other is unfeasible. Various contrasting studies have suggested that an observed correlation indicates good preservation (Bryant et al., 1996; Iacumin et al., 1996a, 1996b), while divergence from a strict correlation is still observed in some well-preserved samples (Martin et al., 2008; Kirsanow & Tuross, 2011; Pellegrini et al., 2011). Given that this study incorporated external controls for diagenesis and included only well-preserved samples in all subsequent correlation calculations, we hereby contend that presence/absence of correlation between $\delta^{18}O_{structural\ carbonate}$ and $\delta^{18}O_{phosphate}$ is not indicative of diagenetic state. Given that this study produced different correlations in different body elements (i.e. enamel, dentin, or bone), it appears that this method of examining diagenesis is unreliable, and we recommend using it with caution. Consideration must be given to the body element tested if such a correlation is used as a diagenetic indicator. Rather it is hereby recommended that an alternate method of examining diagenesis be used, such as collagen quality indicators or examination of recrystallization (Person et al., 1995; Kohn et al., 1999; Thompson et al., 2011). All such methods have their own inherent level of error, but with proper consideration of the data are likely to provide a better indication of diagenesis than isotope comparisons alone.

Conclusion

This study determined human-specific isotope spacing values and regression equations for the relationships between $\delta^{13}C_{collagen,}$ $\delta^{13}C_{structural\ carbonate,}$ $\delta^{18}O_{structural\ carbonate,}$ and $\delta^{18}O_{phosphate}$ in individuals with a largely terrestrial diet. The average $\Delta^{13}C_{carb-coll}$ of 5.4% is fairly consistent between bone and dentin with a strong correlation between $\delta^{13}C_{collagen}$ and $\delta^{13}C_{structural\ carbonate}$. The lack of correlation between $\Delta^{13}C_{carb-coll}$ and $\delta^{15}N$ suggests that these individuals allocate a similar proportion of animal protein to their carbon input regardless of absolute meat intake. The average $\Delta^{18}O_{carb-phos}$ of 7.8% is lower than previous studies, but is consistent between bone, dentin, and enamel. Enamel shows the strongest correlation between $\delta^{18}O_{\text{structural carbonate}}$ and $\delta^{18}O_{\text{phosphate}}$ values compared to dentin and bone. This has potential implications concerning isotopic fractionation of blood-DIC during collagen formation, as well as potential implications for species-specific fractionations of oxygen during mineralization. Using the $\delta^{18}O_{structural\ carbonate}$ and correlation for determination

diagenetic state is tenuous and should be used with caution and attention to the particular body element (i.e. bone, dentin, enamel) being considered. This robust data set can now provide a basis of comparison for human isotope studies which typically involve small sample numbers and an attempt to limit destructive analyses of valuable and rare specimens.

Acknowledgements

Karin Bruwelheide and Aleithea Warmack helped procure samples, and Charlotte Doney, Sara McGuire, Sara Mills, and Whitney Miller assisted with specimen preparation. Collection access was authorized by Lisa Croft, Mark Hungerford, and Jeffrey Hanson, Bureau of Reclamation, Albuquerque, NM; Rev. Larry Madden, Holy Trinity Catholic Church, Georgetown, D.C.; Nicholas Bellantoni, Connecticut State Archaeologist, Storrs, CT: Franklin Damann and Brian Spatola, the National Museum of Health and Medicine, Silver Spring, MD; Chuck Fithian, the Division of Historical and Cultural Affairs, State of Delaware, Dover, DE; Yvonne Oakes, the Museum of New Mexico, Santa Fe, NM; the Association for the Preservation of Historic Congressional Cemetery, Washington, DC, and David Hunt, National Museum of Natural History, Smithsonian Institution, Washington, DC.

References

Ambrose SH. 1990. Preparation and characterization of bone and tooth collagen for isotopic analysis. *Journal of Archaeological Science* 17: 431–451.

Ambrose SH. 1993. Isotopic Analysis of paleodiets: methodological and interpretive considerations. In *Investigations of Ancient Human Tissue: Chemical Analysis in Anthropology*, MK Sanford (ed.). Gordon Breach Science Publishers: Langhorne, USA; 59–130.

Ambrose SH, Norr L. 1993. Experimental evidence for the relationship of the carbon isotope ratios of whole diet and dietary protein to those of bone collagen and carbonate. In *Prehistoric Human Bone — Archaeology at the Molecular Level*, JB Lambert, G Grupe (eds.). Springer-Verlag: Berlin; 1–33.

Balasse M, Bocherens H, Mariotti A. 1999. Intra-bone variability of collagen and apatite isotopic composition used as evidence of a change of diet. *Journal of Archaeological Science* **26**: 593–598.

Billings SA, Richter DD. 2006. Changes in stable isotopic signatures of soil nitrogen and carbon during 40 years of forest development. *Oecologia* 148: 325–333.

- Bocherens H, Drucker D. 2003. Trophic level isotopic enrichment of carbon and nitrogen in bone collagen: case studies from recent and ancient terrestrial ecosystems. *International Journal of Osteoarchaeology* **13**: 46–53.
- Bocherens H, Billiou D, Patou-Mathis M, Bonjean D, Otte M, Mariotti A. 1997. Paleobiological implications of the isotopic signatures (¹³C, ¹⁵N) of fossil mammal collagen in Scladina Cave (Sclayn, Belgium). *Quaternary Research* **48**: 370–380.
- Bocherens H, Fizet M, Mariotti A. 1994. Diet, physiology and ecology of fossil mammals as inferred from stable carbon and nitrogen isotope biogeochemistry: implications for Pleistocene bears. *Palaeogeography*, *Palaeoclimatology*, *Palaeoecology* 107: 213–225.
- Bocherens H, Fizet M, Mariotti A, Lange-Badre B, Vandermeersch B, Borel JP, Bellon G. 1991. Isotopic biogeochemistry ¹³C, ¹⁵N of fossil vertebrate collagen: application to the study of a past food web including Neandertal man. *Journal of Human Evolution* **20**: 481–492.
- Bocherens HM, Pacaud G, Lazarev PA, Mariotti A. 1996. Stable isotope abundances (¹³C, ¹⁵N) in collagen and soft tissues from Pleistocene mammals from Yakutia: implications for the palaeobiology of the Mammoth Steppe. *Palaeogeography*, *Palaeoclimatology*, *Palaeoecology* **126**: 31–44.
- Bryant JD, Froelich PN. 1995. A model of oxygen isotope fractionation in body water of large mammals. *Geochimica et Cosmochimica Acta* **59**: 4523–4537.
- Bryant JD, Koch PL, Froelich PN, Showers WJ, Genna BJ. 1996. Oxygen isotope partitioning between phosphate and carbonate in mammalian apatite. *Geochimica et Cosmochimica Acta* 60: 5145–5148.
- Coltrain JB. 2009. Sealing, whaling, and caribou revisited: additional insights from the skeletal isotope chemistry of eastern Arctic foragers. *Journal of Archaeological Science* 36: 764–775
- Coltrain JB, Harris JM, Cerling TE, Ehleringer JR, Dearing M, Ward J, Allen J. 2004. Rancho La Brea stable isotope biogeochemistry and its implications for the palaeoecology of the late Pleistocene, coastal southern California. *Palaeogeography*, *Palaeoclimatology*, *Palaeoecology* 205: 199–219.
- D'Angela D, Longinelli A. 1990. Oxygen isotopes in living mammal's bone phosphate: further results. *Chemical Geology* **86**: 75–82.
- Daux V, Lécuyer C, Héron M, Amiot R, Simon L, Fourel F, Martineau F, Lynnerup N, Reychler H, Escarguel G. 2008. Oxygen isotope fractionation between human phosphate and water revisted. *Journal of Human Evolution* 55: 1138–1147.
- DeNiro MJ. 1985. Postmortem preservation and alteration of in vivo bone collagen isotope ratios in relation to palaeodietary reconstruction. *Nature* **317**: 806–809.
- DeNiro MJ, Epstein S. 1978. Influence of diet on the distribution of carbon isotopes in animals. *Geochimica et Cosmochimica Acta* 42: 495–506.
- DeNiro MJ, Epstein S. 1981. Influence of diet on the distribution of nitrogen isotopes in animals. *Geochimica et Cosmochimica Acta* 45: 341–351.

- Dettman DL, Kohn MJ, Quade J, Ryerson FJ, Ojha TP, Hamidullah S. 2001. Seasonal stable isotope evidence for a strong Asian monsoon throughout the past 10.7 m. y. Geology 29: 31–34.
- Drucker D, Bocherens H, Bridault A, Billiou D. 2003. Carbon and nitrogen isotopic composition of red deer (*Cervus elaphus*) collagen as a tool for tracking palaeoenvironmental change during the Late-Glacial and Early Holocene in the northern Jura (France). *Palaeogeography*, *Palaeoclimatology*, *Palaeoecology* **195**: 375–388.
- Drucker D, Bocherens H, Pike-Tay A, Mariotti A. 2001. Isotopic tracking of seasonal dietary change in dentine collagen: preliminary data from modern caribou. *Earth and Planetary Sciences* **333**: 303–309.
- Fox-Dobbs K, Bump JK, Peterson RO, Fox DL, Koch PL. 2007. Carnivore-specific stable isotope variables and variation in the foraging ecology of modern and ancient wolf populations: case studies from Isle Royale, Minnesota, and La Brea. *Canadian Journal of Zoology* 85: 458–471.
- Francillon-Vieillot H, de Buffrénil V, Castanet J, Géraudie J, Meunier FJ, Sire JY, Zylberberg L, de Ricqlès A. 1990. Microstructure and mineralization of vertebrate skeletal tissues. In *Skeletal Biomineralization: Patterns, Processes, and Evolutionary Trends Volume I, JG Carter* (ed.). Van Norstrand Reinhold: New York, USA; 471–530.
- Froehle AW, Kellner CM, Schoeninger MJ. 2010. FOCUS: effect of diet and protein source on carbon stable isotope ratios in collagen: follow up to Warinner and Tuross (2009). *Journal of Archaeological Science* 37: 2662–2670.
- Handley LL, Austin AT, Robinson D, Scrimgeour CM, Raven JA, Heaton THE, Schmidt S, Stewart GR. 1999. The ¹⁵N abundance (δ^{15} N) of ecosystem samples reflects measures of water availability. *Australian Journal of Plant Physiology* **26**: 185–199.
- Heaton THE. 1999. Spatial, species, and temporal variations in the $^{13}\text{C}/^{12}\text{C}$ ratios of C_3 plants: implications for palaeodiet studies. *Journal of Archaeological Science* **26**: 637–649.
- Hedges REM. 2003. On bone collagen-apatite-carbonate isotopic relationships. *International Journal of Osteoarchaeology* **13**: 66–79.
- Hilderbrand GV, Farley SD, Robbins CT, Hanley TA, Titus K, Servheen C. 1996. Use of stable isotopes to determine diets of living and extinct bears. Canadian Journal of Zoology 74: 2080–2088.
- Howland MR, Corr LT, Young SMM, Jones V, Jim S, van der Merwe NJ, Mitchell AD, Evershed RP. 2003. Expression of the dietary isotope signal in the compound-specific δ^{13} C values of pig bone lipids and amino acids. *International Journal of Osteoarchaeology* **13**: 54–65.
- Iacumin P, Bocherens H, Mariotti A, Longinelli A. 1996a. An isotopic palaeoenvironmental study of human skeletal remains from the Nile Valley. Palaeogeography, Palaeoclimatology, Palaeocology 126: 15–30.
- Iacumin P, Bocherens H, Mariotti A, Longinelli A. 1996b. Oxygen isotope analyses of co-existing carbonate and phosphate in biogenic apatite: a way to monitor

- diagenetic alteration of bone phosphate? Earth and Planetary Science Letters 142: 1–6.
- Jim S, Ambrose SH, Evershed RP. 2004. Stable carbon isotopic evidence for differences in the dietary origin of bone cholesterol, collagen, and apatite: implications for their use in palaeodietary reconstruction. Geochimica et Cosmochimica Acta 68: 61–72.
- Jorkov MLS, Heinemeier J, Lynnerup N. 2007. Evaluating bone collagen extraction methods for stable isotope analysis in dietary studies. *Journal of Archaeological Science* 34: 1824–1829.
- Kirsanow K, Tuross N. 2011. Oxygen and hydrogen isotopes in rodent tissues: impact of diet, water, and ontogeny. *Palaeogeography*, *Palaeoclimatology*, *Palaeoecology* 310: 9–16.
- Koch PL. 1998. Isotopic reconstruction of past continental environments. Annual Reviews in Earth and Planetary Science 26: 573–613.
- Koch PL, Fogel ML, Tuross N. 1994. Tracing the diets of fossil animals using stable isotopes. In Stable Isotopes in Ecology and Environmental Science, K Lajtha, RH Michener (eds.). Blackwell Scientific Publications: London; 63–92.
- Kohn MJ. 1996. Predicting animal δ^{18} O: accounting for diet and physiological adaptation. *Geochimica et Cosmochimica Acta* **60**: 4811–4829.
- Kohn MJ, Cerling TE. 2002. Stable isotope compositions of biological apatite. In Phosphates: Geochemical, Geobiological, and Materials Importance, reviews in Mineralogy and Geochemistry vol. 48, MJ Kohn, J Rakovain, JM Hughes (eds.). The Mineralogical Society of America: Washington, 455–480.
- Kohn MJ, Schoeninger MJ, Berker WM. 1999. Altered states: Effects of diagenesis on fossil tooth chemistry. Geochimica et Cosmochimica Acta 63: 2737–2747.
- Krueger HW, Sullivan GH. 1984. Models for carbon isotope fractionation between diet and bone. In *Stable Isotopes in Nutrition ACS Symposium Series* 258, JF Turnland, PE Johnson (eds.). American Chemical Society: Washington; 205–222.
- Lee-Thorp JA, Sealy JC, van der Merwe NJ. 1989. Stable carbon isotope ratio differences between bone collagen and bone apatite, and their relationship to diet. *Journal of Archaeological Science* **16**: 585–599.
- Levinson AA, Luz B, Kolodny Y. 1987. Variations in oxygen isotopic compositions of human teeth and urinary stones. *Applied Geochemistry* **2**: 367–371.
- Loftus E, Sealy J. 2012. Technical note: interpreting stable carbon isotopes in human tooth enamel: an examination of tissue spacing from South Africa. *American Journal of Physical Anthropology* **147**: 499–507.
- Longin R. 1971. New method of collagen extraction for radiocarbon dating. *Nature* **230**: 241–242.
- Longinelli A. 1984. Oxygen isotopes in mammal bone phosphate: A new tool for paleohydrological and paleoclimatological research? Geochimica et Cosmochimica Acta 48: 385–390.
- Luz B, Kolodny Y. 1985. Oxygen isotope variations in phosphate of biogenic apatites, IV. Mammal teeth and bones. *Earth and Planetary Science Letters* **75**: 29–36.

- Luz B, Kolodny Y, Horowitz M. 1984. Fractionation of oxygen isotopes between mammalian bone-phosphate and environmental drinking water. *Geochimica et Cosmochimica Acta* 48: 1689–1693.
- MacFadden BJ. 2000. Cenozoic mammalian herbivores from the Americas: Reconstructing ancient diets and terrestrial communities. *Annual Review of Ecology and Systematics* 31: 33–59
- Martin C, Bentaleb I, Kaandorp R, Iacumin P, Chatri K. 2008. Intra-tooth study of modern rhinoceros enamel δ^{18} O: is the difference between phosphate and carbonate δ^{18} O a sound diagenetic test? *Palaeogeography*, *Palaeoclimatology*, *Palaeoecology* **266**: 183–189.
- McNulty T, Calkins A, Ostrom P, Ganghi H, Gottfried M, Martin L, Gage D. 2002. Stable isotope values of bone organic matter: artificial diagenesis experiments and paleoecology of Natural Trap Cave, Wyoming. *Palaios* 17: 36–49.
- van der Merwe NJ. 1982. Carbon isotopes, photosynthesis, and archaeology. *American Scientist* **70**: 596–606.
- Michel V, Ildefonse P, Morin G. 1995. Chemical and structural changes in *Cervus elaphus* tooth enamels during fossilization (Lazaret cave): a combined IR and XRD Rietvold analysis. *Applied Geochemistry* 10: 145–159.
- Minagawa M, Wada E. 1984. Stepwise enrichment of ^{15}N along food chains: further evidence and the relation between $\delta^{15}N$ and animal age. Geochimica et Cosmochimica Acta 48: 1135–1140.
- Nadelhoffer KJ, Colman BP, Currie WS, Magill A, Aber JD. 2004. Decadal-scale fates of ¹⁵N tracers added to oak and pine stands under ambient and elevated N inputs at the Harvard Forest (USA). Forest Ecology and Management 196: 89–107
- Nelson BK, De Niro MJ, Schoeninger MJ, DePaolo DJ, Hare PE. 1986. Effects of diagenesis on strontium, carbon, nitrogen, and oxygen concentration and isotopic composition of bone. *Geochimica et Cosmochimica Acta* **50**: 1941–1949.
- O'Leary MH. 1988. Carbon isotopes in photosynthesis. *BioScience* **38**: 328–336.
- Pardo LH, McNulty SG, Boggs JL, Duke S. 2007. Regional patterns in foliar ¹⁵N across a gradient of nitrogen deposition in the northeastern US. *Environmental Pollution* **149**: 293–302.
- Passey BH, Robinson TF, Ayliffe LK, Cerling TE, Sponheimer M, Dearing MD, Roeder BL, Ehleringer JR. 2005. Carbon isotope fractionation between diet, breath CO₂, and bioapatite in different mammals. *Journal of Archaeological Science* **32**: 1459–1470.
- Pellegrini M, Lee-Thorp JA, Donahue RE. 2011. Exploring the variation of the $\delta^{18}O_p$ and $\delta^{18}O_c$ relationship in enamel increments. *Palaeogeography*, *Palaeoclimatology*, *Palaeoecology* 310: 71–83.
- Person A, Bocherens H, Mariotti A, Renard M. 1996. Diagenetic evolution and experimental heating of bone phosphate. *Palaeogeography*, *Palaeoclimatology*, *Palaeoecology* 126: 135–149.

- Person A, Bocherens H, Saliège J, Paris F, Zeitoun V, Gérard M. 1995. Early diagenetic evolution of bone phosphate: An X-ray diffractometry analysis. *Journal of Archaeological Science* 22: 211–221.
- Peterson BJ, Fry B. 1987. Stable isotopes in ecosystem studies. Annual Review of Ecology and Systematics 18: 293–320.
- Post DM. 2002. Using stable isotopes to estimate trophic position: models, methods, and assumptions. *Ecology* 83: 703–718.
- Roth JD, Hobson KA. 2000. Stable carbon and nitrogen isotopic fractionation between diet and tissue of captive red fox: implications for dietary reconstruction. *Canadian Journal of Zoology* 78: 848–852.
- Schimmelmann A, Albertino A, Sauer PE, Qi H, Molinie R, Mesnard F. 2009. Nicotine, acetanilide and urea multilevel ²H, ¹³C, and ¹⁵N-abundance reference materials for continuous flow isotope ratio mass spectrometry. *Rapid Communications in Mass Spectrometry* 23: 3513–3521.
- Schoeninger MJ, DeNiro MJ. 1984. Nitrogen and carbon isotopic composition of bone collagen from marine and terrestrial animals. *Geochimica et Cosmochimica Acta* 48: 625–639.
- Smith BN, Epstein S. 1971. Two categories of ¹³C/¹²C ratios for higher plants. *Plant Physiology* 47: 380–384.
- Sutoh M, Koyama T, Yoneyama T. 1987. Variations of natural ¹⁵N abundances in the tissues and digesta of domestic animals. *Radioisotopes* **36**: 74–77.
- Templer PH, Arthur MA, Lovett GM, Weathers KC. 2007. Plant and soil natural abundance δ¹⁵N: indicators of relative rates of nitrogen cycling in temperate forest ecosystems. *Oecologia* **153**: 399–406.
- Thompson TJU, Islam M, Piduru K, Marcel A. 2011. An investigation into the internal and external variable acting on crystallinity index using Fourier Transform Infrared

- Spectroscopy on unaltered and burned bone. *Palaeogeography*, *Palaeoclimatology*, *Palaeoecology* **299**: 168–174.
- Tieszen LL, Fagre T. 1993. Effect of diet quality and composition on the isotopic composition of respiratory CO₂, bone collagen, bioapatite, and soft tissues. In *Prehistoric Human Bone Archaeology at the Molecular Level*, JB Lambert, G Grupe (eds.). Springer-Verlag: Berlin; 121–155.
- Tuross N, Behrensmeyer AK, Eanes ED, Fisher LW, Hare PE. 1989. Molecular preservation and crystallographic alterations in a weathering sequence of wildebeest bones. *Applied Geochemistry* **4**: 261–270.
- Tütken T, Vennemann TW, Pfretzschner HU. 2008. Early diagenesis of bone and tooth apatite in fluvial and marine settings. Constraints from combined oxygen isotope, nitrogen and REE analysis. *Palaeogeography*, *Palaeoclimatology*, *Palaeoecology* **266**: 254–268.
- Veis A. 2003. Mineralization in organic matrix frameworks. In Reviews in Mineralogy and Geochemistry, volume 54: Biomineralization, PM Dove, JJ DeYoreo, S Weiner (eds.). The Mineralogical Society of America: Washington; 249–289.
- Warinner C, Tuross N. 2010. Brief Communication: Tissue isotopic enrichment associated with growth depression in a pig: implications for archaeology and ecology. *American Journal of Physical Anthropology* 141: 486–493.
- Zazzo A, Balasse M, Passey BH, Moloney AP, Monahan FJ, Schmidt O. 2010. The isotope record of short- and long-term dietary changes in sheep tooth enamel: implications for quantitative reconstruction of paleodiets. *Geochimica et Cosmochimica Acta* 74: 371–3586.
- Zazzo A, Lécuyer C, Sheppard SMF, Grandjean P, Mariotti A. 2004. Diagenesis and the reconstruction of paleoenvironments: a method to restore original δ^{18} O values of carbonate and phosphate from fossil tooth enamel. *Geochimica et Cosmochimica Acta* **68**: 2245–2258.